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Technology Assessment of Artificial Photosynthesis

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CO₂-WIN Virtual Conference | 9th June 2021

Introduction

Where we are now



Where we came from



Components became smaller while devices became larger on purpose

Introduction

Where we are now



Where we came from



Size and price have continuously dropped.

Introduction

Where we are now



Where we came from



Introduction

Where we are now



Where we came from



Digitalization – miniaturization – integration.

If it is conceivable, it is achievable.

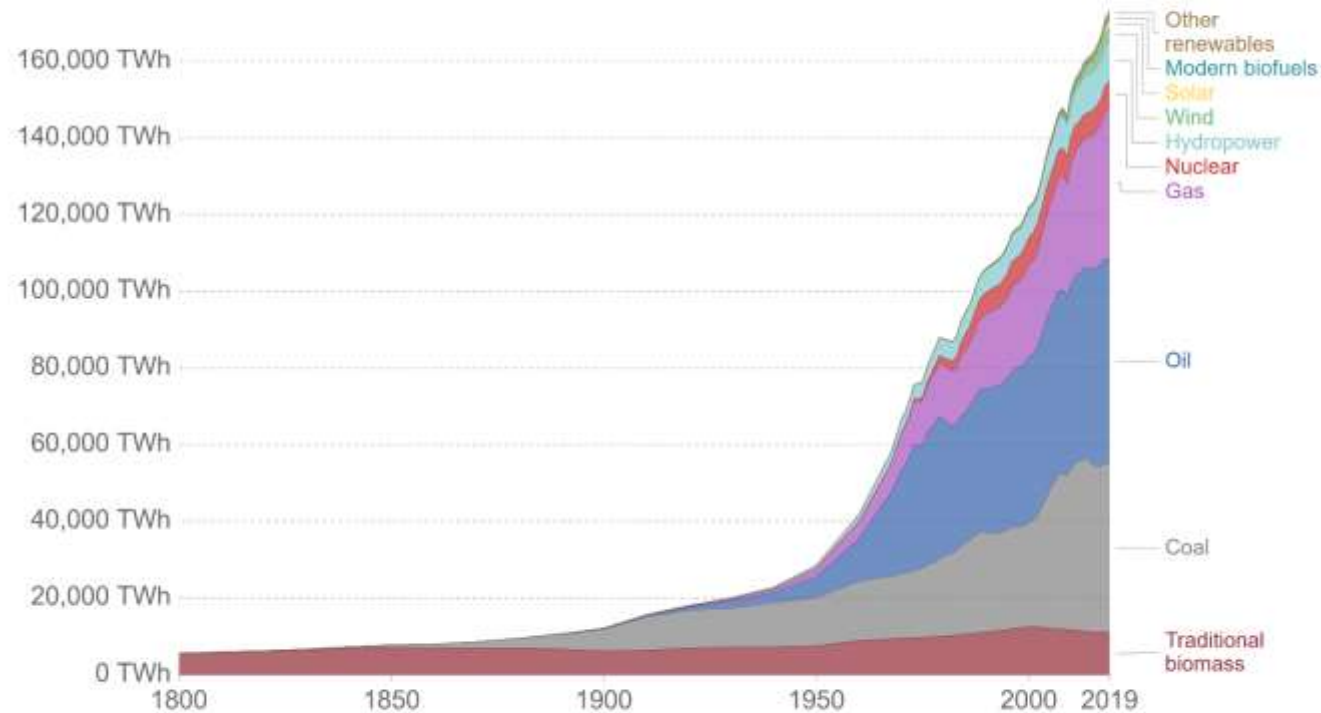
Constant improvement and disruptive innovation will be equally necessary.

Humanity's energy demand vs. sunlight irradiation

Global primary energy consumption by source

Primary energy is calculated based on the 'substitution method' which takes account of the inefficiencies in fossil fuel production by converting non-fossil energy into the energy inputs required if they had the same conversion losses as fossil fuels.

Our World
in Data



Source: Vaclav Smil (2017) & BP Statistical Review of World Energy

OurWorldInData.org/energy • CC BY

[1] <https://ourworldindata.org/energy-production-consumption>

[2] <https://www.sciencedirect.com/topics/engineering/solar-energy>

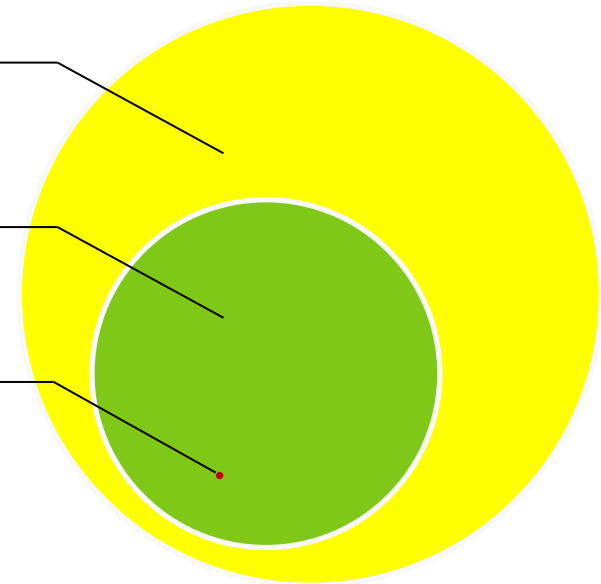
[3] „Power Generation Technologies“ (Chapter 13 – Solar Power), 2019, P. Breeze, <https://www.sciencedirect.com/science/article/pii/B9780081026311000134>

[4] „Detailed Balance Limit of Efficiency of p-n Junction Solar Cells“, J. Appl. Phys. 32, 510, 1961, W. Shockley & H. J. Queisser, <https://aip.scitation.org/doi/10.1063/1.1736034>

[5] „Künstliche Photosynthese Besser als die Natur?“, 2019, H. Dau, P. Kurz, M.-D. Weitze (ISBN: 978-3-662-55718-1)

Humanity's energy demand vs. sunlight irradiation

- Sun energy reaching the surface of the earth per year: 944,444,400 TWh ^[2,3]
- Maximum usable sun energy after Shockley-Queisser-Limit ^[4,5]: 331,000,000 TWh
- Energy demand per year: 160,000 TWh ^[1]



- Even considering Shockley-Queisser efficiency limit of $\approx 35\%$, the usable sun energy is 20.000 times higher than our yearly energy demand.
- Hence, 0.34% of land area and 17.5% efficiency would suffice to cover our energy demand.

[1] <https://ourworldindata.org/energy-production-consumption>

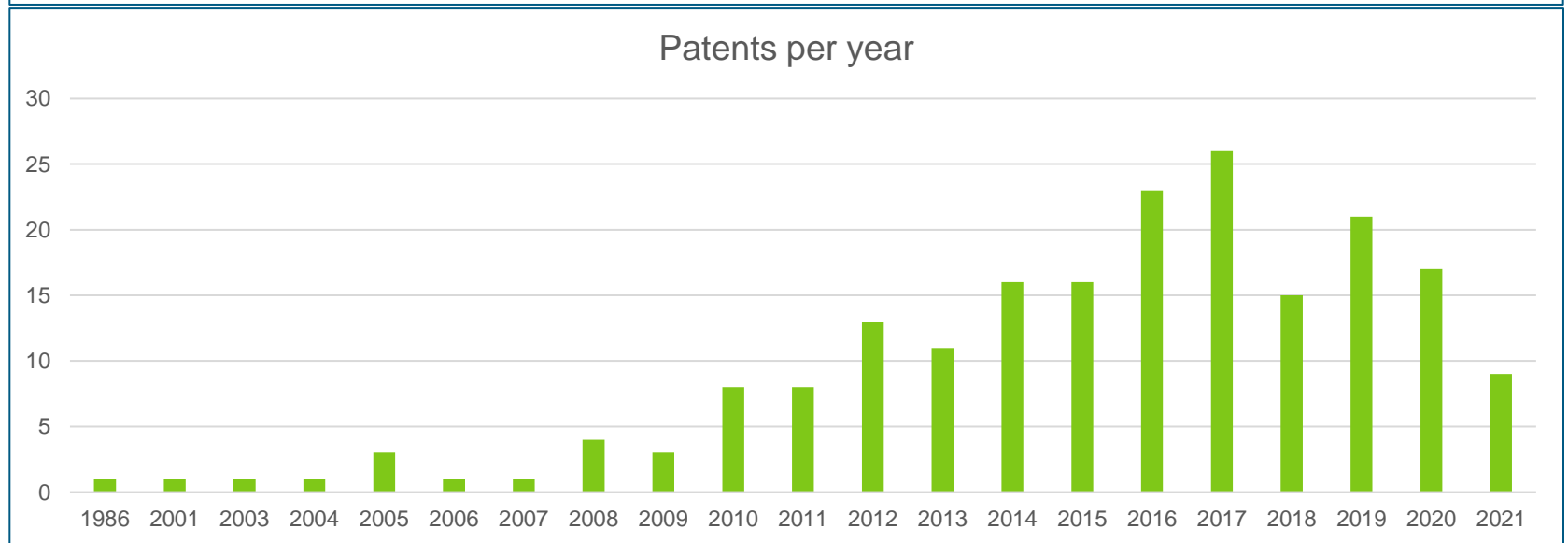
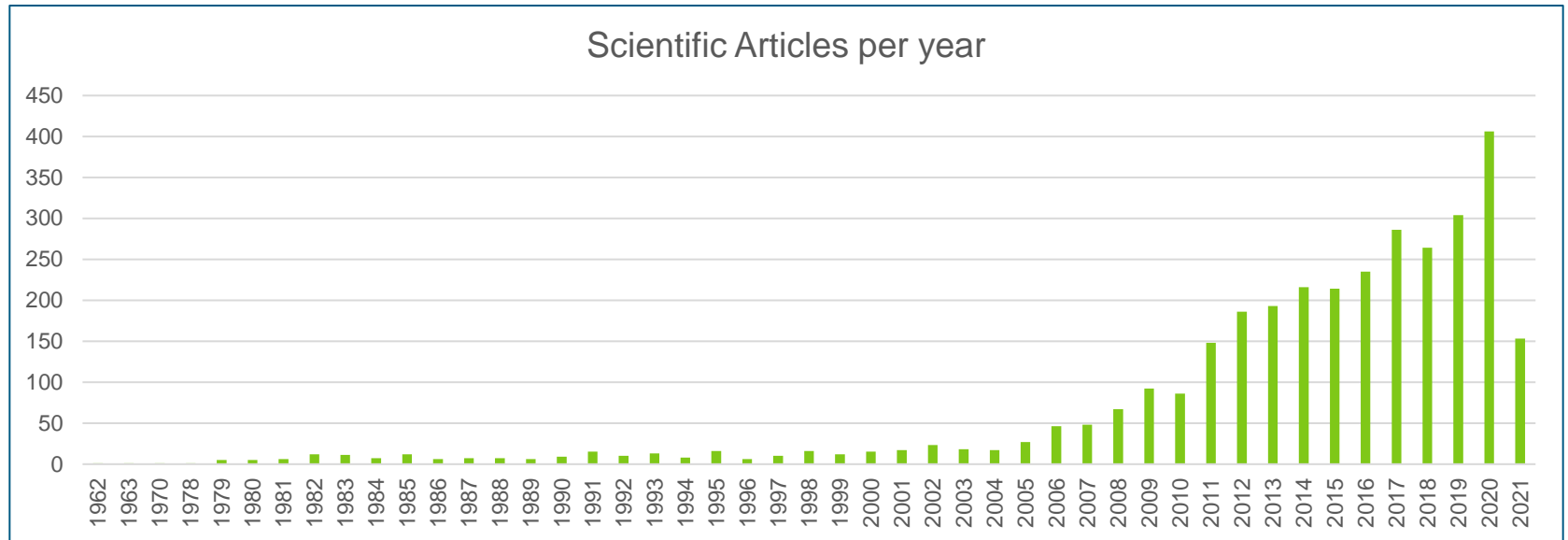
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[4] „Detailed Balance Limit of Efficiency of p-n Junction Solar Cells“, W. Shockley & H. J. Queisser, J. Appl. Phys. 32, 510, 1961, <https://aip.scitation.org/doi/10.1063/1.1736034>

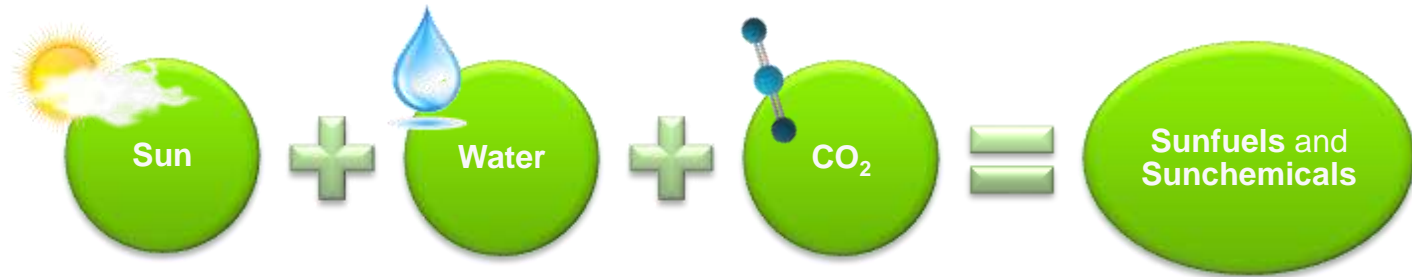
[5] „Künstliche Photosynthese Besser als die Natur?“, H. Dau, P. Kurz, M.-D. Weitze, 2019 (ISBN: 978-3-662-55718-1)

Patents Charts



Reference: SciFinder query for the phrase „artificial photosynthesis“ – analysis by type = „patent“, May 30, 2021

Concepts of Artificial Photosynthesis



Artificial Photosynthesis Routes

Technologic

Photoelectro-
chemical

Photochemical

Hybrid

Photobio-
electrochemical

Biologic (mod.)

Enzymatic (cell-free)

Full Cell Catalysis

Achievements toward solar Hydrogen production

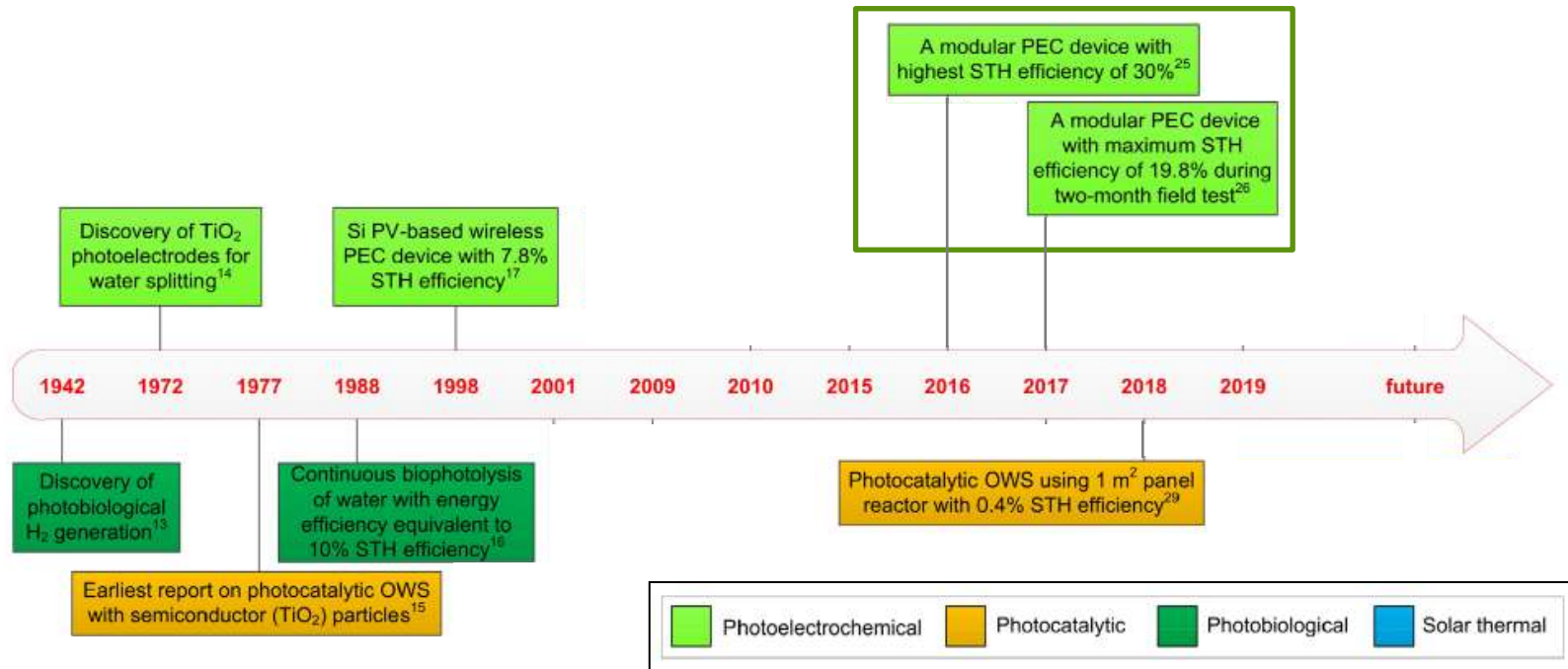


Image adapted from: „Research advances towards large-scale solar hydrogen production from water”, G. Liu, Y. Sheng, J. W. Ager, M. Kraft, R. Xu, EnergyChem 1(2), 2019, 100014, <https://doi.org/10.1016/j.enchem.2019.100014>

Solar Hydrogen production benchmarks – PV plus electrolysis

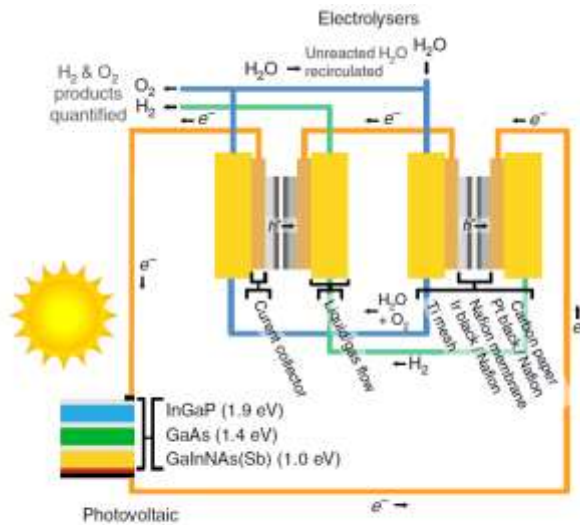


Figure 1 | PV-electrolysis device schematic. The PV-electrolysis system consists of a triple-junction solar cell and two PEM electrolyzers connected in series.

Product	H ₂ (+O ₂) (separated)
STH	30%
Materials	<ul style="list-style-type: none"> • 3J solar cell (InGaP, GaAs, GaInNAs(Sb)) • Two PEM electrolyzers
Lifetime	> 2 d

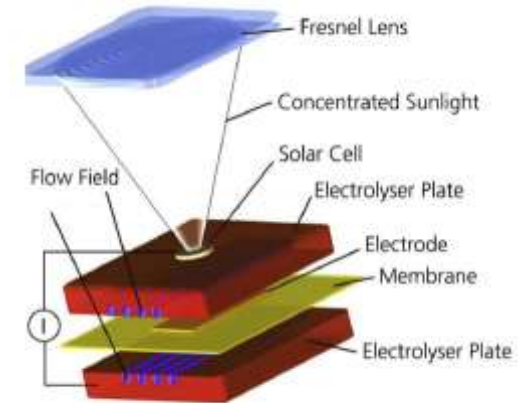


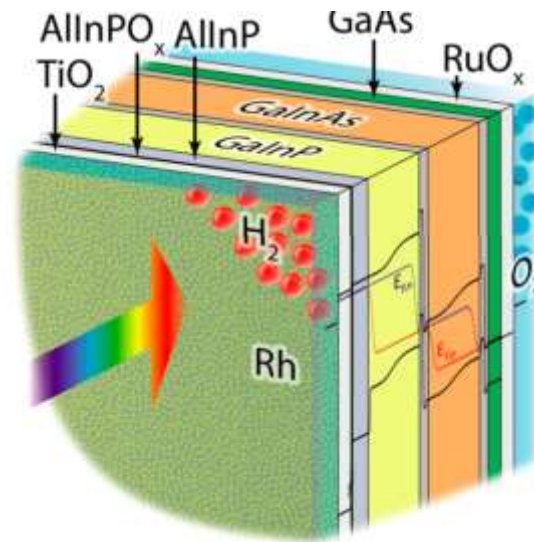
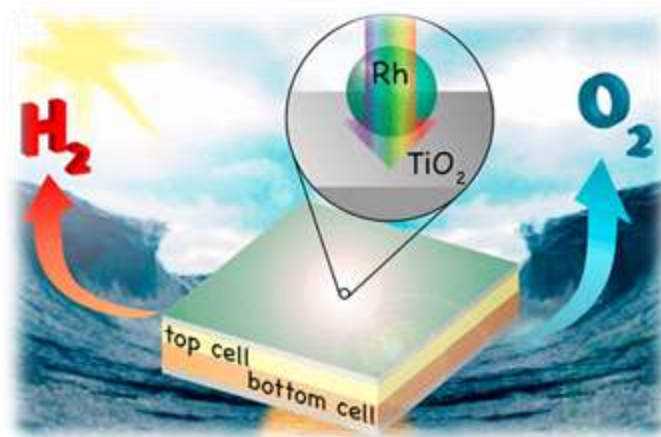
Fig. 1 – Schematic view of the HyCon concept [13]. A

Product	H ₂ (+O ₂) (separated)
STH	20%
Materials	<ul style="list-style-type: none"> • Solar cell not specified. • Electrolyzer not specified.
Lifetime	> 60 d

Left: „Solar water splitting by photovoltaic-electrolysis with a solar-to-hydrogen efficiency over 30%”, J. Jia, L. C. Seitz, J. D. Benck, Y. Chen, J. W. D. Ng, T. Bilir, J. S. Harris, Th. F. Jaramillo, Nature Communications 7, 2016, <https://doi.org/10.1038/ncomms13237>

Right: “Hydrogen concentrator demonstrator module with 19.8% solar-to-hydrogen conversion efficiency according to the higher heating value”, A. Fallisch, L. Schellhase, J. Fresko, M. Zedda, J. Ohlmann, M. Steiner, A. Bösch, L. Zielke, S. Thiele, F. Dimroth, T. Smolinka, Int. J. Hydrogen Energy 43(4), 2017, 26804, <https://doi.org/10.1016/j.ijhydene.2017.07.069>

Solar Hydrogen production – Artificial Leaf – integrated system



Product	H ₂ (+O ₂) (separated)
STH	18-19% (at AM 1.5G)
Materials	<ul style="list-style-type: none"> • RuO₂ (for OER-Electrode) • GaInP/GaInAs on GaAs substrate (Photoelectrode) TiO₂ (anatase) layer (by ALD) • Rh (Nanoparticles)
Lifetime	≈ 1 d („stability remains an issue“)

Image Source: „Monolithic Photoelectrochemical Device for Direct Water Splitting with 19% Efficiency”, W.-H. Cheng, M. H. Richter, M. M. May, J. Ohlmann, D. Lackner, F. Dimroth, Th. Hannappel, H. A. Atwater, H.-J. Lewerenz, ACS Energy Lett. 3(8), 2018, 1795-1800, <https://doi.org/10.1021/acsenerylett.8b00920>

C-compounds production from solar energy

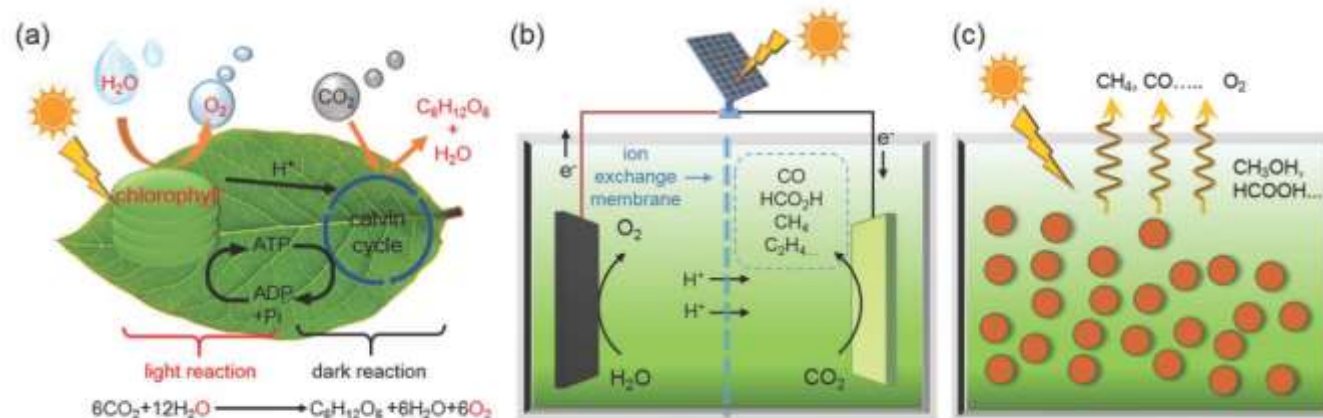


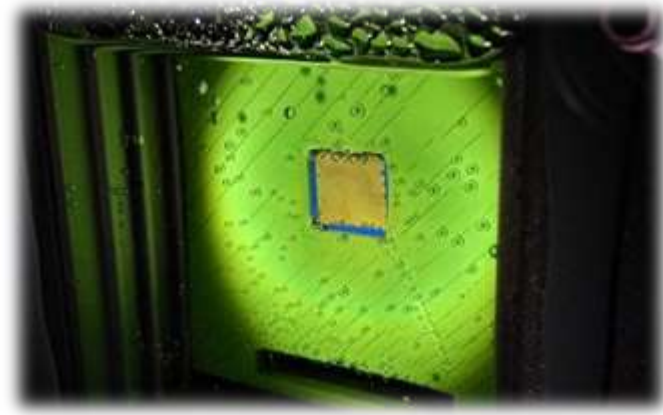
Figure 2. Analogy among a) natural photosynthesis, b) electrochemical synthesis on electrocatalysts powered by a photovoltaic cell, and c) photochemical synthesis on powdery photocatalysts.

- Very different reaction conditions (catalysts, co-catalyst, light sources, ...) to produce different compounds.
- Achievable products: CO, Methane, Methanol, Formic Acid, Ethane, Ethanol

Recent publications toward photocatalytic C-compounds production



Product	HCOOH (+O ₂)
STP	0.08% (at 98% selectivity)
Materials	<ul style="list-style-type: none"> • SrTiO₃:La,Rh Au • RuO₂-BiVO₄:Mo Au • CotpyP (immobilized Cobalt-phosphoryl complex)
Lifetime	50% of initial reaction rate after 24 h. (recovered to 80% after reloading CotpyP)



Product	H ₂ + CO (= syngas)
STH	0.06 %
ST-CO	0.02 %
Materials	<ul style="list-style-type: none"> • Catalyst: Cobalt porphyrin catalyst immobilized on carbon nanotubes • Photoabsorbers: Triple-cation mixed halide perovskite and BiVO₄
Lifetime	≈ 3 d

Left: „Molecularly engineered photocatalyst sheet for scalable solar formate production from carbon dioxide and water”, Q. Wang, J. Warnan, S. Rodriguez-Jimenwz, J. L. Leung, S. Kalathil, V. Andrei, K. Domen, E. Reisner, Nature Energy 5, 2020, 703-710, <https://doi.org/10.1038/s41560-020-0678-6>

Right: “Bias-free solar syngas production by integrating a molecular cobalt catalyst with perovskite–BiVO₄ tandems”, V. Andrei, B. Reuillard, E. Reisner, Nature Materials 19, 2020, <https://doi.org/10.1038/s41563-019-0501-6>

Advantages & Challenges of different routes

Route	Advantages	Challenges
Photo-electro-chemical	<ul style="list-style-type: none">• High efficiencies.	<ul style="list-style-type: none">• Increase system integration and decrease interfacial losses.• Increase lifetime.• Decrease costs.
Photo-catalytic	<ul style="list-style-type: none">• Inexpensive.• Robust.• Simple.	<ul style="list-style-type: none">• Increase efficiencies.
Biologic	<ul style="list-style-type: none">• General concepts are well-developed.	<ul style="list-style-type: none">• Sensitive to reaction conditions and environment.

TRL assessment

TRL 3: Experimental proof of concept

- ✓ First laboratory scale prototype (proof-of-concept) or numerical model realized
- ✓ Testing at laboratory level of the innovative technological element (being material, sub-component, software tool, ...), but not the whole integrated system
- ✓ Key parameters characterizing the technology (or the fuel) are identified
- Verification of experimental application through simulation tools and cross-validation with literature data (if applicable).

TRL 4: Technology validated in lab

- 🕒 (Reduced scale) prototype developed and integrated with complementing sub-systems at laboratory level
- Validation of the new technology through enhanced numerical analysis (if applicable).
- ✓ Key Performance Indicators are measurable
- 🕒 The prototype shows repeatable/stable performance (either TRL4 or TRL5, depending on the technology)

Based on the TRL definitions from: „Technology Readiness Level: Guidance Principles for Renewable Energy technologies“ (Annexes), European Commission (DG RTD), 2017, <https://op.europa.eu/de/publication-detail/-/publication/d5d8e9c8-e6d3-11e7-9749-01aa75ed71a1>

Economic Assessment – A look into the future



From lab... ...to something that looks like this... ...to a business case.

What do we know?

- ✓ Possible products that can be produced by APS
- ✓ Market size, production costs and price of the products for today

Where we need assumptions

- ? Materials used for a market ready technology → CapEx
- ? Market situation in the future (20 - 30 years) → OpEx, serviceable market and selling price

Economic Assessment – Example Hydrogen

Hydrogen is expected to cover almost one fifth of global demand for final energy in 2050.

Today	Production volume	Production costs per ton	
Steam reforming	117 Mt, Global, 2019	≈ 1.400 €	
Electrolysis	Negligible	3.000 – 5.500 €	

Future needs for APS	Lifetime	Solar to Hydrogen efficiency	Area
Artificial Photosynthesis	10 years	10%	≈ 58.500 km ²

Current and future needs

- To speak a common language, i.e., develop a common understanding of knowledge and challenges.
 - Open research data, e.g., the materials genome.
 - Artificial intelligence and computational methods.
- An intense and multidisciplinary collaboration to make fast and big steps at the same time.
 - National research centers & think tanks
 - International infrastructures and user facilities
- To demonstrate scalable systems with medium (short-term) and high (mid-term) efficiencies and lifetimes.
- To develop standards to ease market penetration.