

Project TRANSFORMATE

Combined process development of electro-chemical CO₂-reduction and synthetic biology for production of the biopolymer PHB and crotonic acid

Frank Kensy, Project Coordinator

1. Status Conference CO₂-WIN

9th June 2021

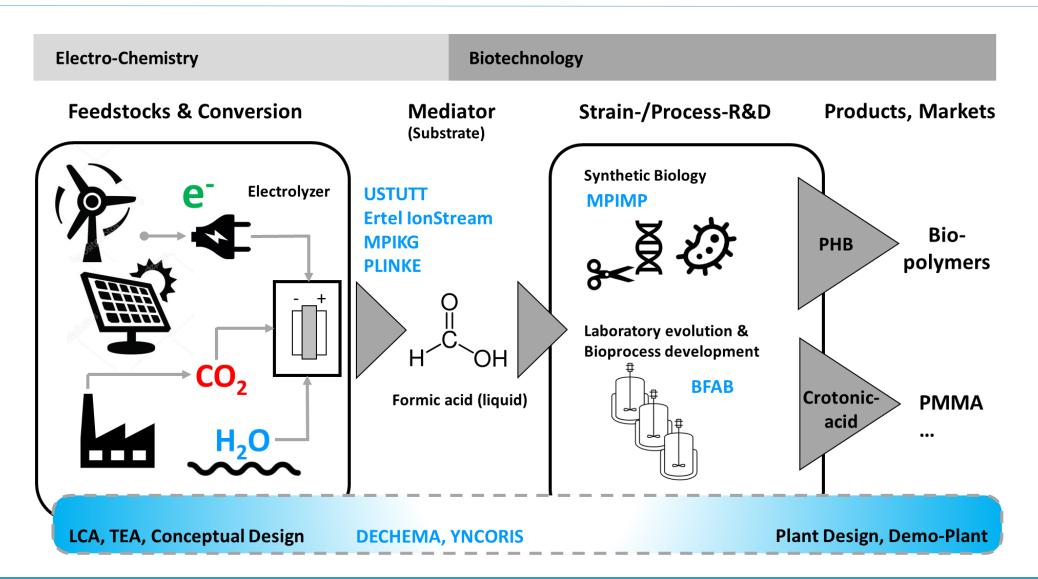
SPONSORED BY THE



Federal Ministry of Education and Research

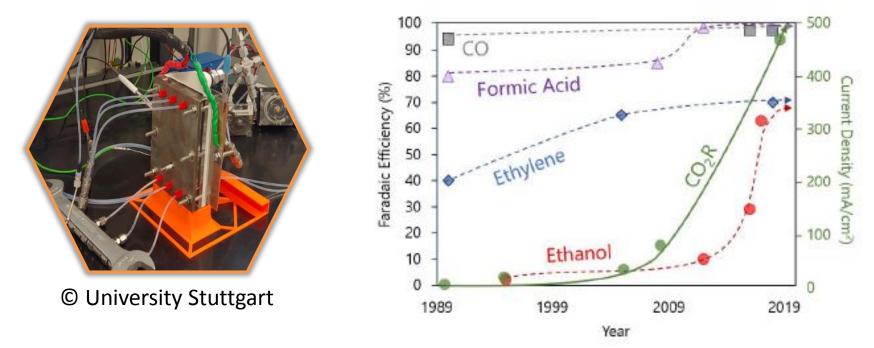
Project Overview





Why CO₂ Reduction to Formic Acid?



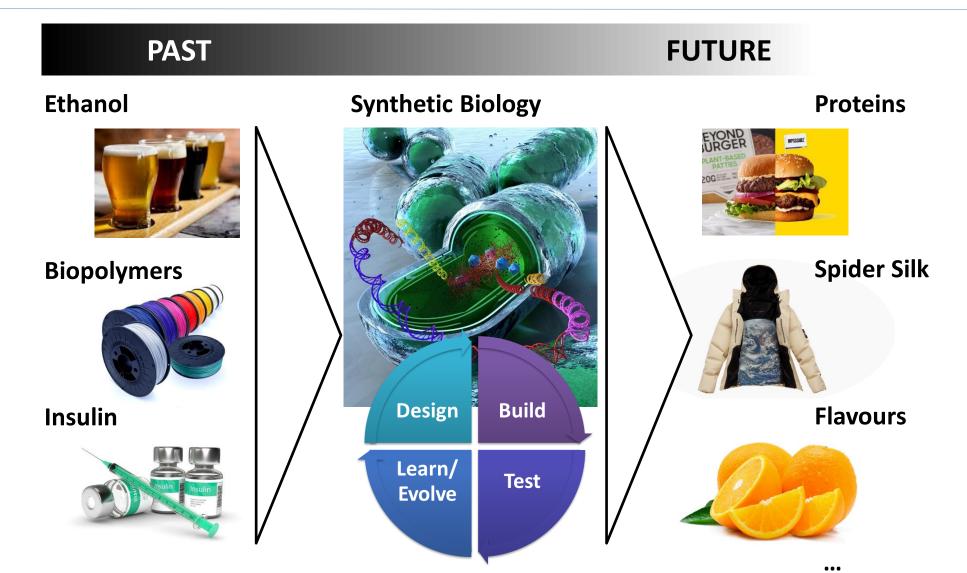


- CO₂ and H₂ storage in a liquid
- High energetic efficiency (FE > 90%, EE > 30-50%)
- High current densities (normally 200-300 mA/cm², max. 1 A/cm² reached)
- No mass transfer limitation in fermentation (fully soluble)

De Luna et al., Science 364, 350 (2019)

Biological Transformation through Synthetic Biology





Goals of the project





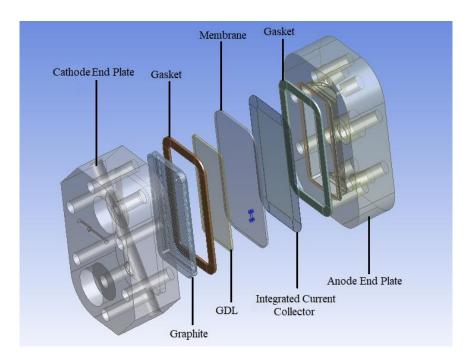
Specific project goals:

- 1. CO₂-electrolyzer for the efficient production of formic acid from CO₂ and water
- 2. Platform strain *Cupriavidus necator* which is capable to growth on formate via rGP
- 3. Production strain and -process for PHB
- 4. Production strain and -process for crotonic acid
- 5. Engineering and plant concept for industrial implementation (including LCA & TEA)

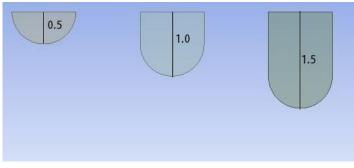
Laboratory CO₂-Electrolyzer



- ➤ Anode
- Cathode
- Flow Field
- Membrane, Gas Diffusion
 Layer and Sealing Gaskets



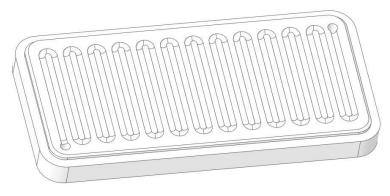
Lab Electrolyzer: explosion view



Flow Field channels with different depth, measured in mm

Manufactured Flow Fields:

- Channel depths, 0.5, 1.0 and 1.5 mm
- Two different graphite grades



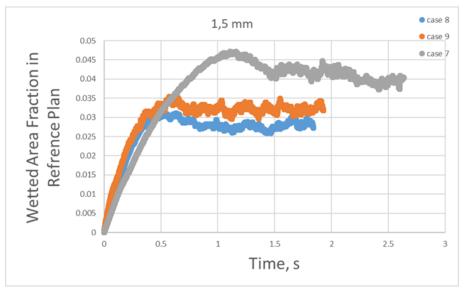
Graphite Flow Field 27 mm × 56 mm

Flow Simulation

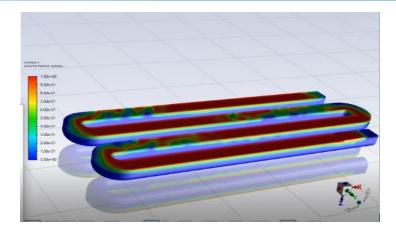


Parameters Studied in Simulation

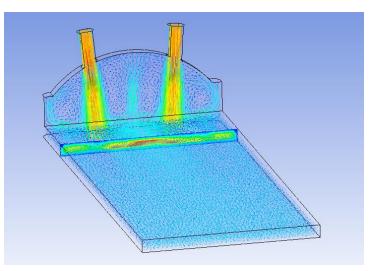
- > Hydrophobicity of the Flow Filed
- Inclination Angle
- Flow Rate of Gas and Liquid in Cathode
- Channel Depth
- Channel Shape
- Anode Inlet/Outlet Manifolds



Simulation Result Analysis



Two-phase Flow Simulation in Channels



Flow Simulation in Anode End-Plate

Confidential | © 2021 TRANSFORMATE, all rights reserved

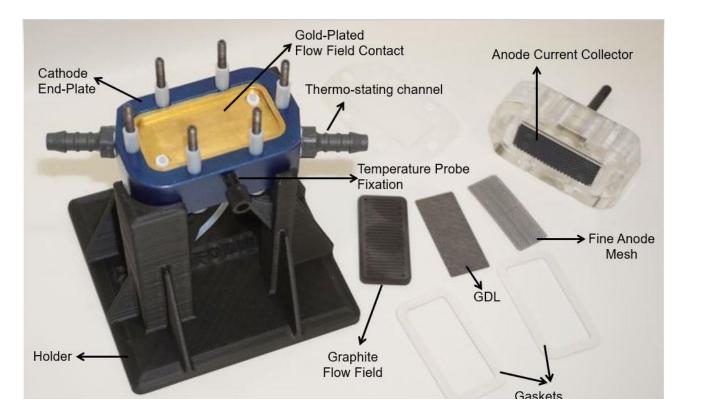
MEA Design with PIL modified Catalytic Layer



Universität Stuttgart



Membrane Electrode Assembly Design (MEA Design) Zero Gap Design



Polymeric Ionic Liquids (PILs)

can be used as...

- ... binders in the CL
- ... stabilisers for metal clusters

in order to

- ... enhance metal dispersion
- ... control ion transport ... pH
 - enhanc
- ... enhance CO₂ flux by selective adsorption

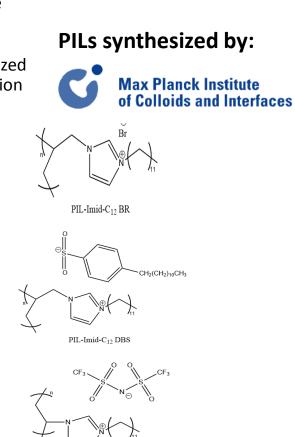
Testing of Binder and Catalyst



Universität Stuttgart

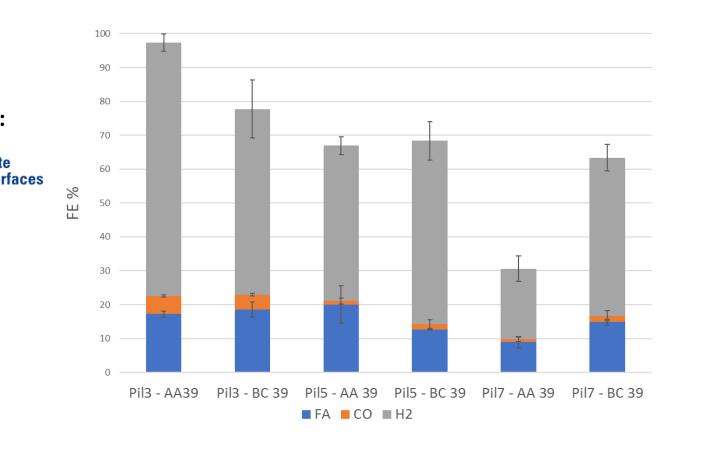


- AA 39 GDL
- BC 39 GDL + MPL
- 35% of PILs for each sample
- SnO2@AB Catalyst
- 1 mg cm-2 of SnO2 synthesized by homogeneous precipitation
- 64% of AB (carbon)
- PIL3: PIL-Imid-C12 Br



PIL7:PIL-Imid-C12 TFSI

• PIL5: PIL-Imid-C12 DBS



GSS analysis conditions: : $j = -200 \text{ mA cm}^{-2}$, $T = 50^{\circ}\text{C}$, 0.5M KHSO₄, pH = 2.5 (H₂SO₄), Ir/Ti anode in H₂O

PIL-Imid-C₁₂ TFSI

Testing of Binder and Catalyst



100

90

80 70

60

40 30

20 10

% 50 **Universität Stuttgart**

PIL3(10%)-PTFE(20%)



PIL3(20%)-PTFE(15%)

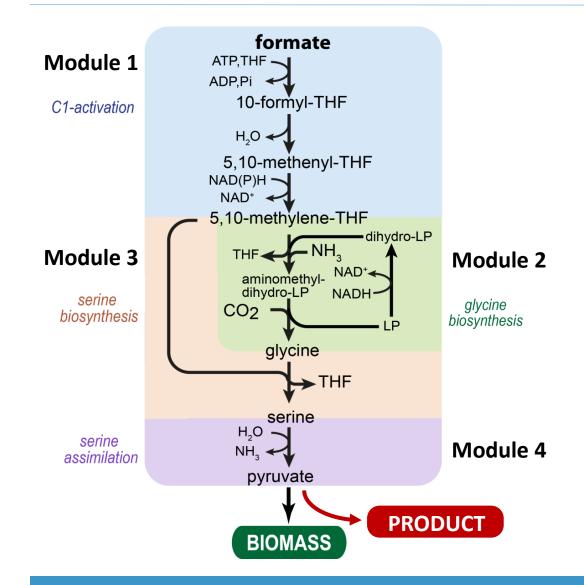
- AA 39 GDL
- J C01 CL: 1 mg cm⁻² of commercial SnO2 supported in AB
- Decreasing amount of PILs increases FE for FA and HER highly suppressed

PIL3(5%)-PTFE(20%)

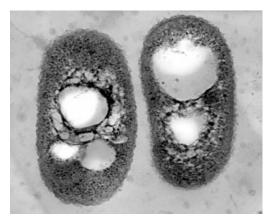
GSS analysis conditions: : $j = -200 \text{ mA cm}^{-2}$, $T = 50^{\circ}\text{C}$, 0.5M KHSO₄, pH = 2.5 (H₂SO₄), Ir/Ti anode in H₂O

Cloning of the Reductive Glycine Pathway





Cupriavidus necator

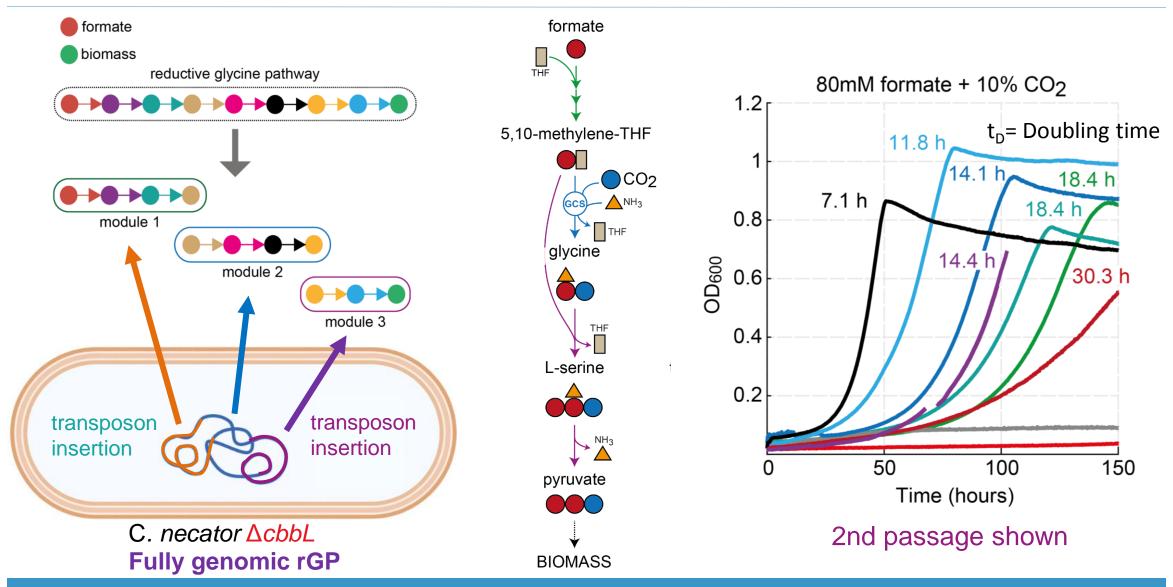


Alternative names: Alcaligenes eutrophus, Ralstonia eutropha, Wautersia eutropha and Hydrogenomonas eutropha

- capable to grow on formate via Calvin Cycle
- natural PHB producer
- versatile host
- basic genetic tools available
- industrial proven microorganism

Genome Integration of 3 Modules of the rGP

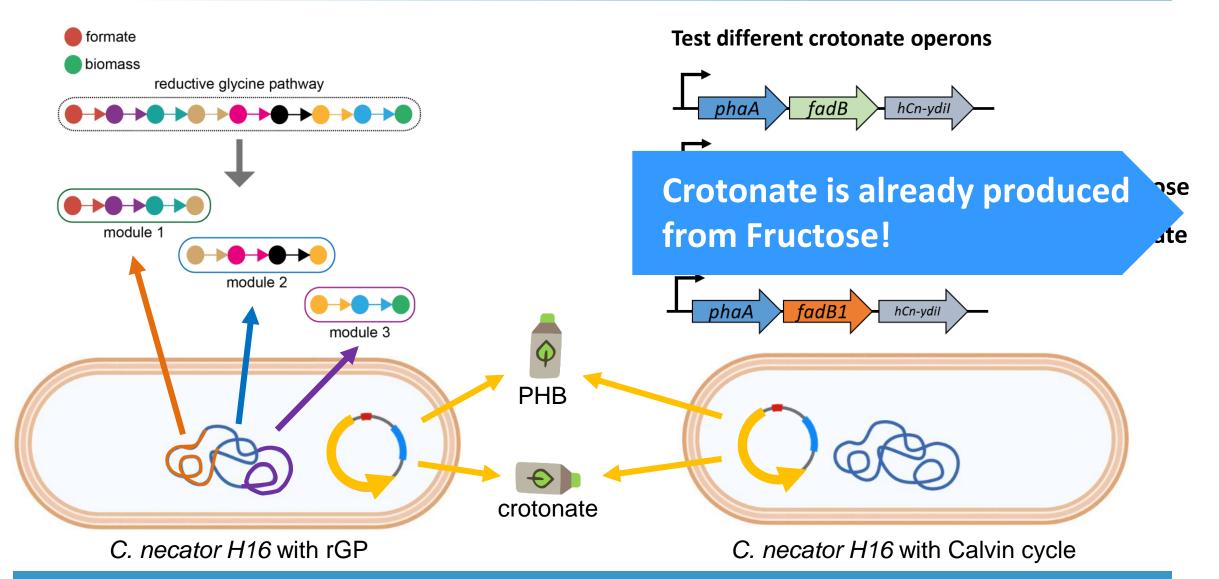
Max-Planck-Institut für Molekulare Pflanzenphysiologie



Confidential | © 2021 TRANSFORMATE, all rights reserved

Product Synthesis via Plug and Play Modules

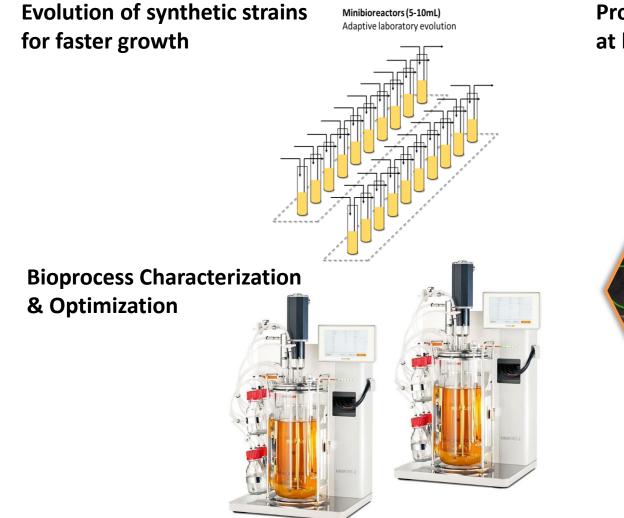




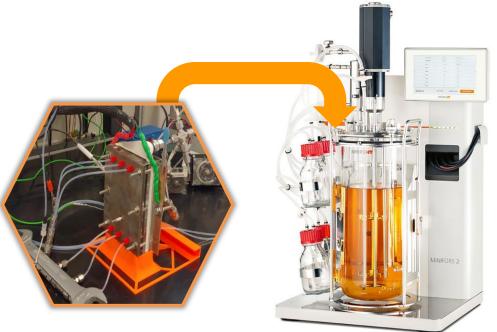
Confidential | © 2021 TRANSFORMATE, all rights reserved

Next steps in Bioprocess Development

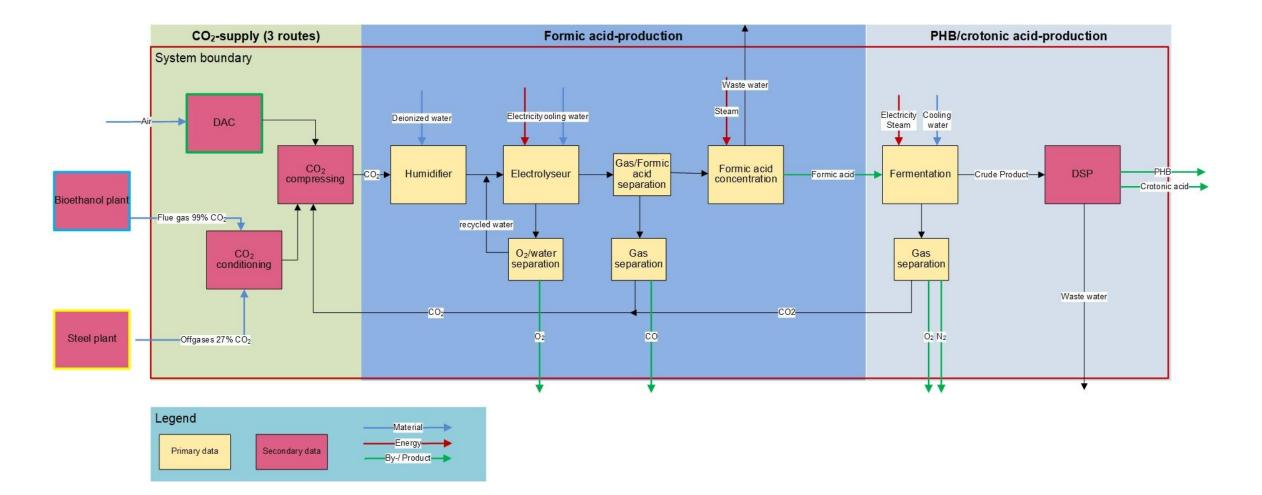




Prozess Integration - Electrolyzer & Bioreactor at laboratory scale (2L)

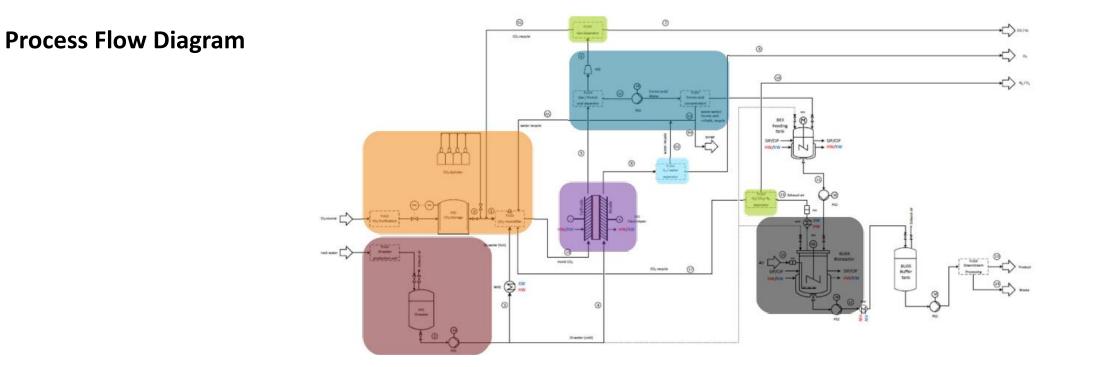






TEA & Conceptual Design





Dashboard - Mass and Energy Balance







- First laboratory CO₂-Electrolyzer built and operated
- 7 different PILs synthesized and tested
- Electrolyzer MEA design is very promising for selective production of formic acid
- Synthetic pathway (rGP) established in *Cupriavidus necator*
- First product (crotonate) synthesized in *Cupriavidus necator*
- LCA model and system boundries fixed
- Process flow diagram and process dashboard established





- Design and construction of a stackable MEA electrolyzer
- Selection of the best PIL for formic acid production
- Optimization and stabilization of electrolyzer performance
- Improvement of cell growth on synthetic pathway via evolution
- Implementation of product synthesis (PHB and crotonate) on new pathway
- Process integration of electrolyzer and bioreactor
- Establishment of a full LCA and TEA



Thank you for your attention!



SPONSORED BY THE

Federal Ministry of Education and Research

Contact: Dr.-Ing. Frank Kensy, kensy@bfab.bio