

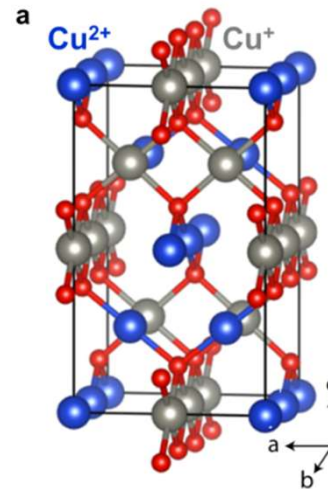
PhasKat

Phase-pure electrocatalysts and adjustment of conditions regarding the reduction of CO_2

CO2- WIN: 1. Status Conference

June 8th, 2021

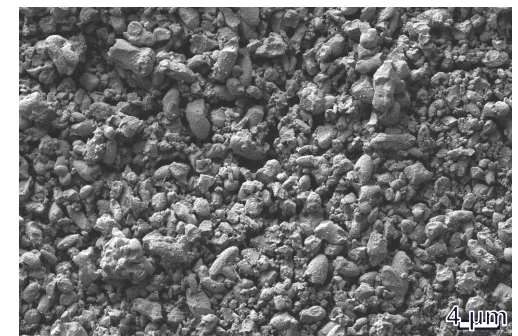
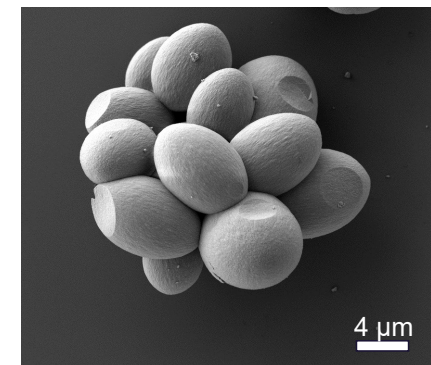
Dr. Wiebke Sarfert-Gast, Siemens Energy Global GmbH & Co. KG



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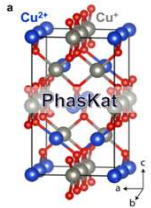


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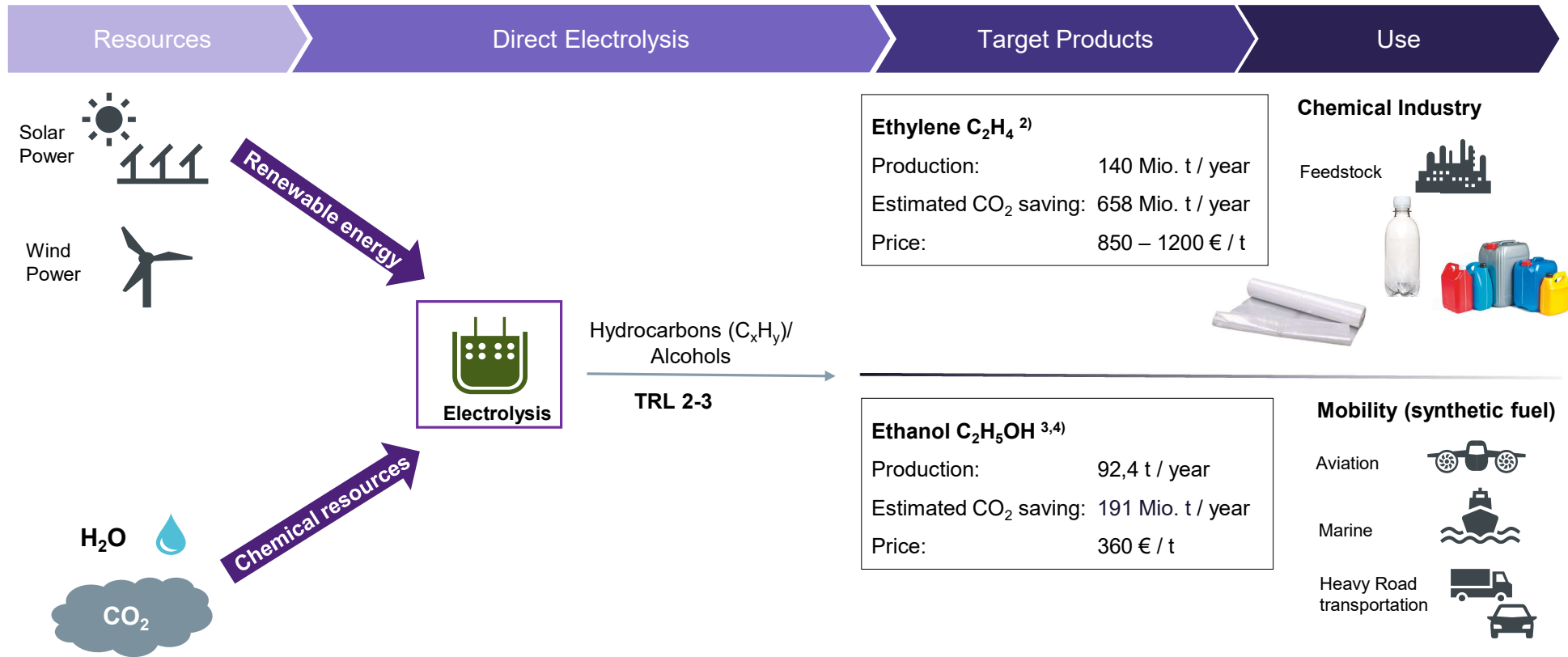
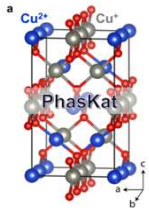


Outline

- Motivation
- Objectives
- Consortium & project setup
- Current status
 - WP1: Catalysts
 - WP2: Anion Exchange Ionomers
 - WP3:
 - Ink preparation & Electrode processing
 - Electrochemical characterization
 - WP4: Characterization by impedance spectroscopy
- Summary & Outlook



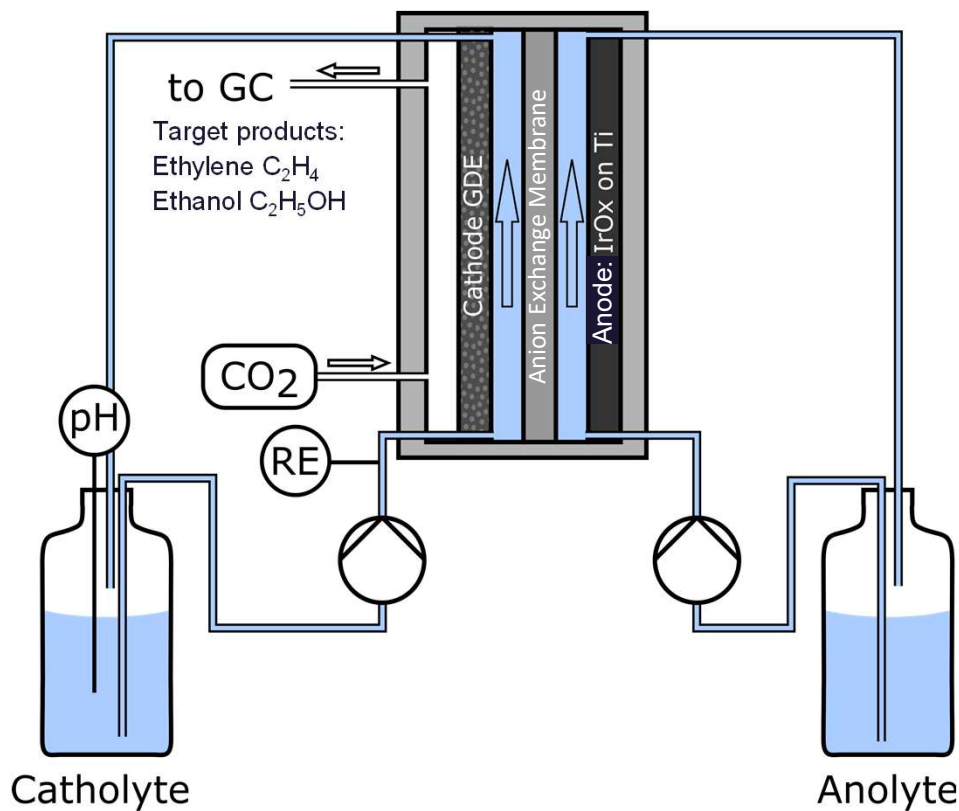
Motivation



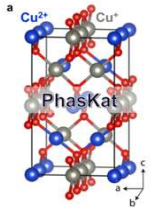
2021-06-08
 1) <http://www.ceresana.com/de/marktstudien/chemikalien/ethylen/>
 2) <https://www.statista.com/statistics/274142/global-ethanol-production-since-2000/>
 3) <https://tradingeconomics.com/commodity/ethanol>

Objective: CO₂-Electrolysis

Gas Diffusion Electrode (GDE) =
Gas diffusion layer coated with catalyst
and Anion Exchange Ionomers (AEI)



2021-06-08



The performance of the GDE depends on:

- Properties of catalyst material
- Properties of Anion Exchange Ionomers
- Layer composition
- Morphology due to processing conditions
- Operation environment
- Operation mode

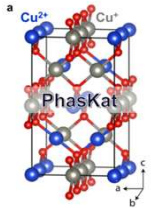
→ The aim of PhasKat is to develop a better understanding about the materials and the way of processing in order to be able to optimize the GDE performance at industrial relevant current densities (> 200 mA/ cm²).

Consortium

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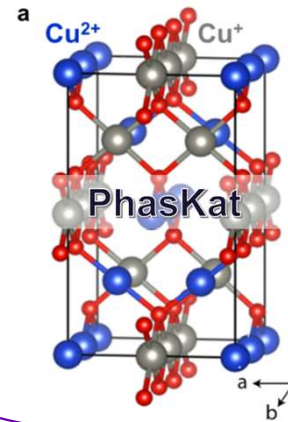
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CREAVIS & Smart Materials

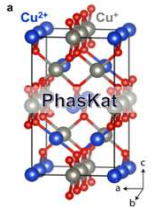


Helmholtz Institute
Erlangen-Nürnberg
part of
Forschungszentrum Jülich



Project Setup

Duration: 01.02.2020 – 31.01.2023
 Funding code: 033RC028



WP1 Catalysts (Evonik SM, SE)

Reference Materials (known to SE)

- Cu_4O_3 (Paramelaconite)
- $\text{Ag}_2\text{Cu}_2\text{O}_3$

New Target Materials

- BaCuO_2
- CuFe_2O_4
- Ag_3CuS_2

WP3 Electrode Processing & Electrochemical Characterization (HI ERN, SE)

- Evaluation of various fabrication methods for GDE preparation.
- Evaluation of the catalysts and AEIs.
- Optimization of GDE (10 cm^2) in terms of performance and stability to meet the final target: $\text{FE}_{\text{C}_2} > 80\% @ 200 \text{ mA/cm}^2, > 100 \text{ h}$.

WP4 GDE Characterization by impedance spectroscopy (FZJ, IEK-9)

- Electrical fundamentals of GDE operation
- Local impedance of single components
- Impedance of electrodes and its interfaces

WP2 Anion Exchange Ionomers (AEI) (Evonik CREAVIS, SE, HIERN)

New AEIs under development:

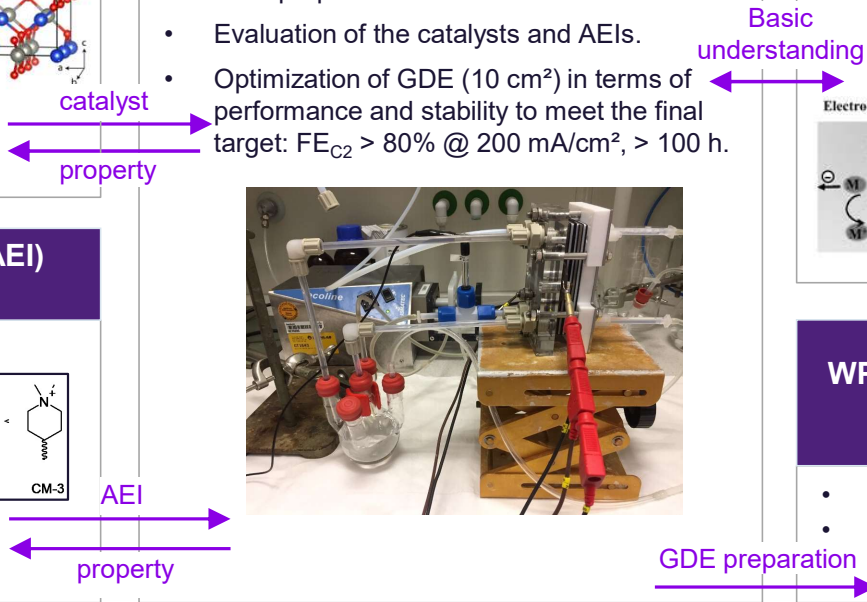
- Evolon

Benchmark AEIs:

- Aemion (Ionomr)
- Sustainion (Dioxide Materials)
-

WP5 System Analysis & Demonstration (Evonik CREAVIS, SE)

- Up-scaling towards 300 cm^2 cell area.
- Demonstration of scalability.



WP1: Catalysts

Objective: Catalyst development, optimization and scale-up



Lab scale preparation



Transfer and implementation of SIEMENS protocol for:

- ✓ Cu_4O_3
- ✓ $\text{Ag}_2\text{Cu}_2\text{O}_3$

Novel Materials:

- ✓ $\text{BaCuO}_{2/2.5}$
- Ag_3CuS_2
- CuFe_2O_4

Project status: ongoing



2021-06-08

Analytics



Transfer of SIEMENS standard analytical protocol for target materials

- Derive analytical tools for novel materials

Project status: implemented



Optimization and scale-up



Constant assessment of recipe and handling optimization as well as scale-up possibilities



Involved partners:

- Dr. A. Reinsdorf, Evonik SM*
- S. Brinkmann, Evonik SM*
- N. Martic, Siemens Energy*
- Dr. G. Schmid, Siemens Energy*

WP1: Catalysts



Catalyst development, optimization and scale-up – Current status

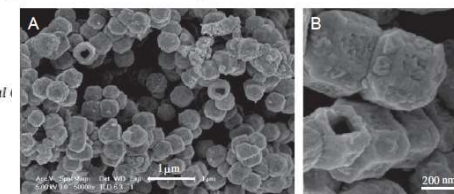
- Intensive literature search was done in close cooperation with SIEMENS
- Different recipes for silver and barium modified copper catalysts were analyzed and are currently in assessment at Evonik Catalysts
- Lab trials started with the preparation of Ag_3CuS_2 in February 2021
- Lab trials with $BaCuO_{2/2.5}$ started in March 2021, two samples are available



Synthesis and ammonia sensing property of Ag_3CuS_2 nanocages obtained from Cu_7S_4 18-facet hollow nanopolyhedra

Zhangxian Chen, Weixin Zhang*, Zeheng Yang

School of Chemical Engineering, Anhui Key Laboratory of Controllable Chemical Reaction and Material



A convenient hydrothermal route to mineral Ag_3CuS_2 nanorods

J.Q. Hu*, B. Deng, W.X. Zhang, K.B. Tang, Y.T. Qian

Structure Research Laboratory and Department of Chemistry, University of Science and Technology of China, Hefei 230026, PR China

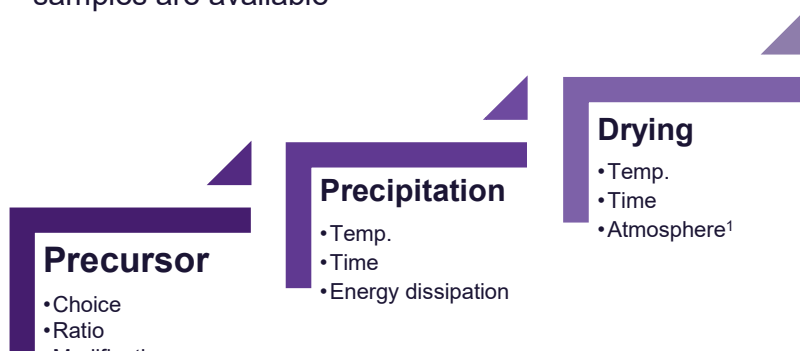
Formation and Decomposition of $BaCuO_{2.5}$ Prepared from a Mixture of Nitrates

MASATO MACHIDA, KIYOSHI YASUOKA, KOICHI EGUCHI, AND HIROMICHI ARAI

Department of Materials Science and Technology, Graduate School of Engineering Sciences, Kyushu University 39, 6-1 Kasugakoen, Kasuga, Fukuoka 816, Japan

Involved partners:

*Dr. A. Reinsdorf, Evonik SM
S. Brinkmann, Evonik SM
N. Martic, Siemens Energy
Dr. G. Schmid, Siemens Energy*



Therm. Treatment

- Temp.
- Time
- Atmosphere¹

¹ Choice of Gases as well as possible convection

WP2: Anion Exchange Ionomer (AEI)

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Objective

- Synthesis of anion exchange ionomers (AEI)

Approach

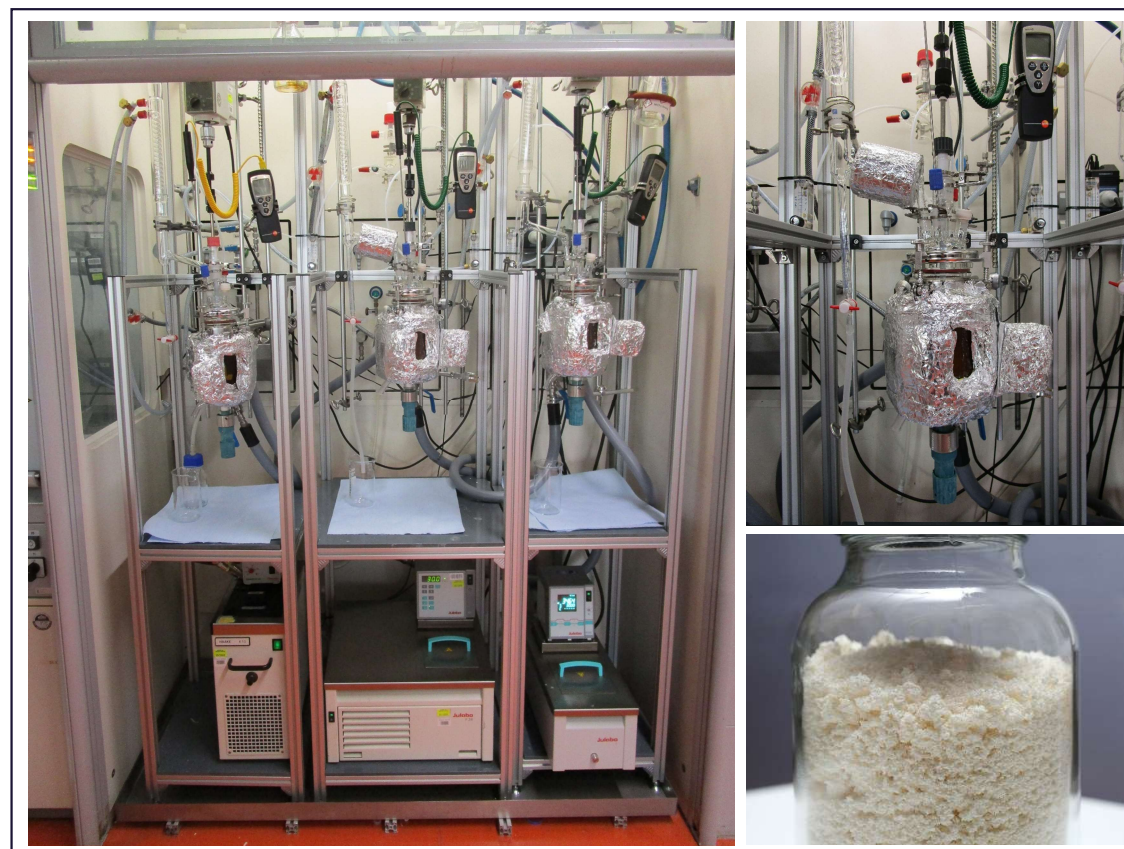
- Synthesis of AEIs using experience of Evonik in development of high-performance polymers

Result

- Evonik provided Siemens samples of Evolon Type I AEI (5g batch)
- Synthesis scaled up to 10g batch

Outlook

- Synthesis of Type I AEI with different IECs & polymerization degree is in progress



Involved partners: Dr. A. Maljusch, Evonik CREAVis

WP3: Catalyst ink composition optimization

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Objective

- Optimizing ink composition to achieve more porous catalyst layer and better performance.

Approach

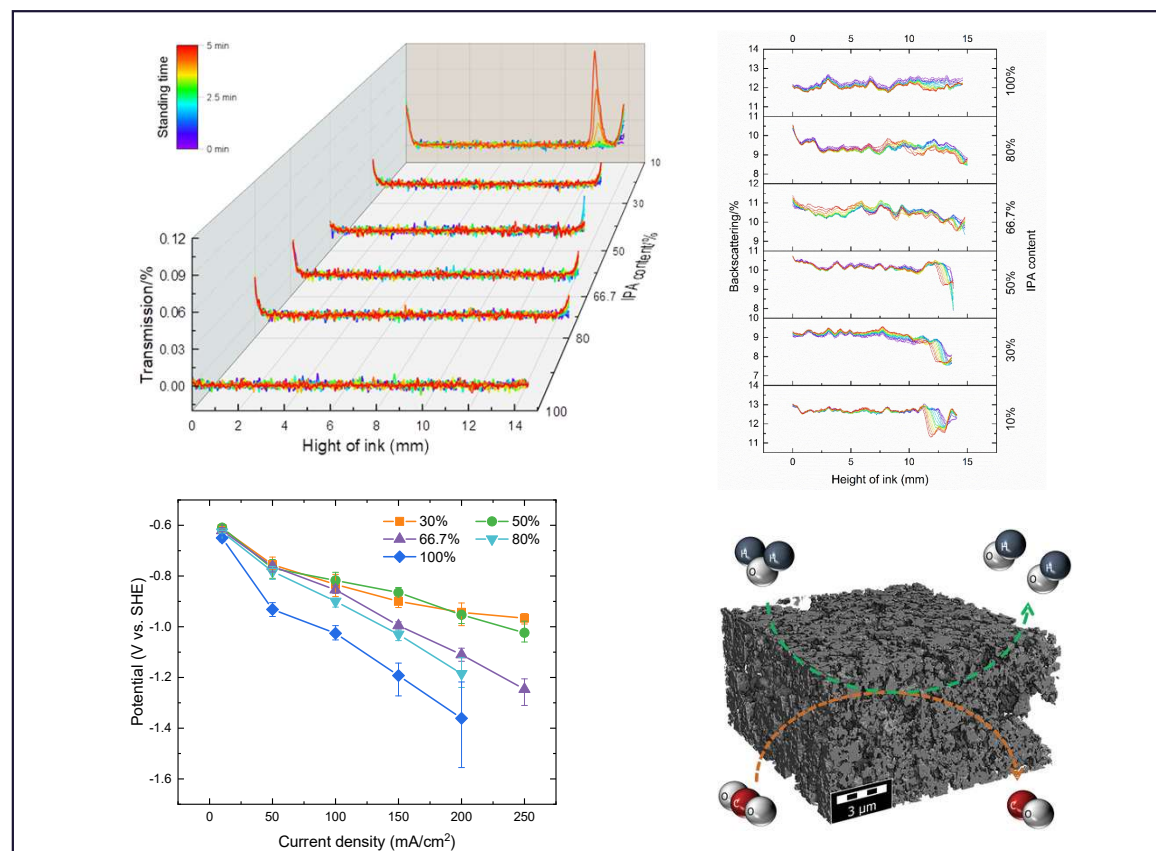
- Adopting an optical method for ink stability analysis, doctor blading for GDE manufacturing and 3D tomographic reconstruction for micro-structure characterization.

Result

- Higher IPA content contributes to stabilize the catalyst ink.
- Lower IPA content results more porous structure and better performance.

Outlook

- Correlating performance with ink composition and catalyst layer structure.



Involved partners: Prof. S. Thiele, L. Guangxin, D. McLaughlin

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WP3: Evaluation of Catalyst Materials

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EVONIK
Leading Beyond Chemistry

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Objective

- Paramelaconite (Cu_4O_3) based catalyst powders provided by Evonik have been evaluated as catalyst materials for electroreduction of CO_2 and compared to reference samples synthesized at SE.

Approach

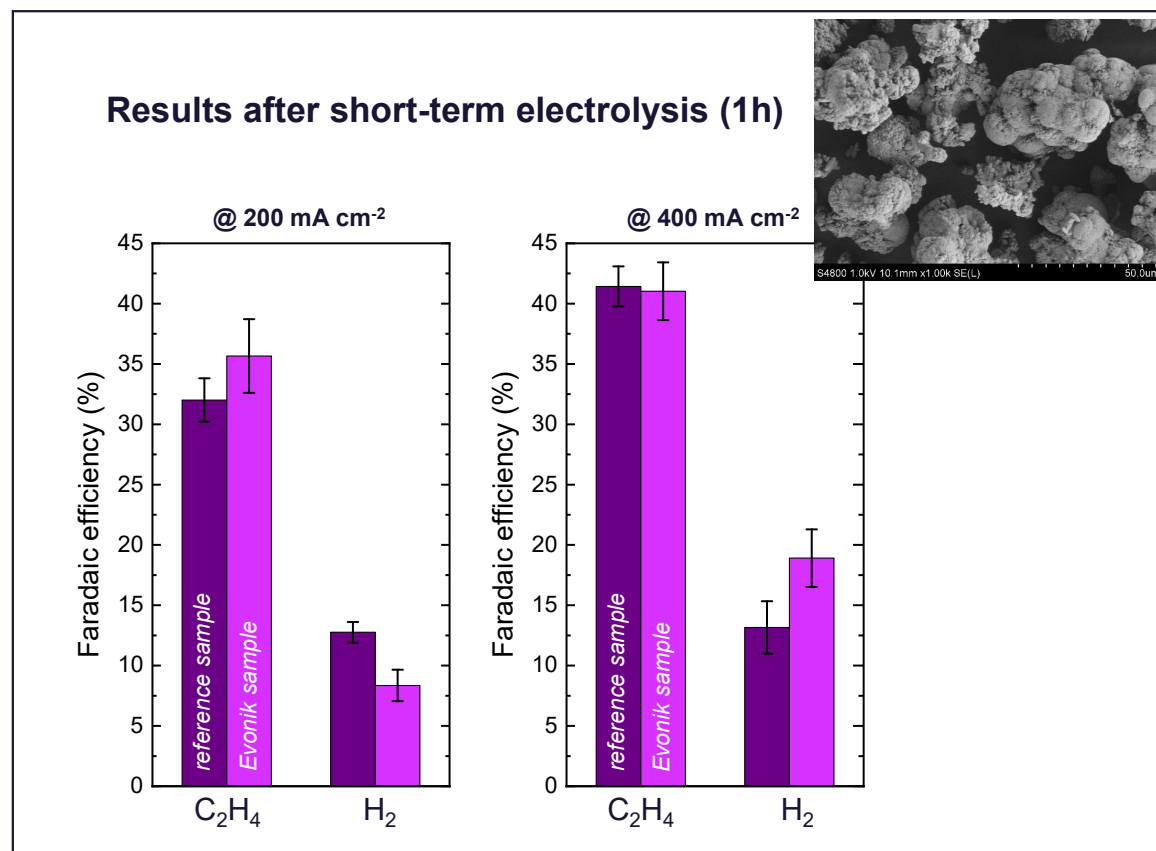
- Investigation via SEM.
- Processing into a GDE using spray-coating.
- Investigation of CO_2 electroreduction properties in a flow cell at industrial relevant current densities ($> 200 \text{ mA/cm}^2$).

Result

- The catalytic performance of Cu_4O_3 powders provided by Evonik are comparable to the reference sample.

Outlook

- Evaluation of catalysts as prepared by Evonik.
- Selection of best catalyst for performance optimization and up-scaling.



Involved partners: Dr. A. Reinsdorf (Evonik); N. Martic, S. Szyszkowski, A. Maltenberger (SE)

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WP3: Evaluation of Anion Exchange Ionomers

Objective

- Evaluation and benchmarking of different anion exchange ionomers (AEI).

Approach

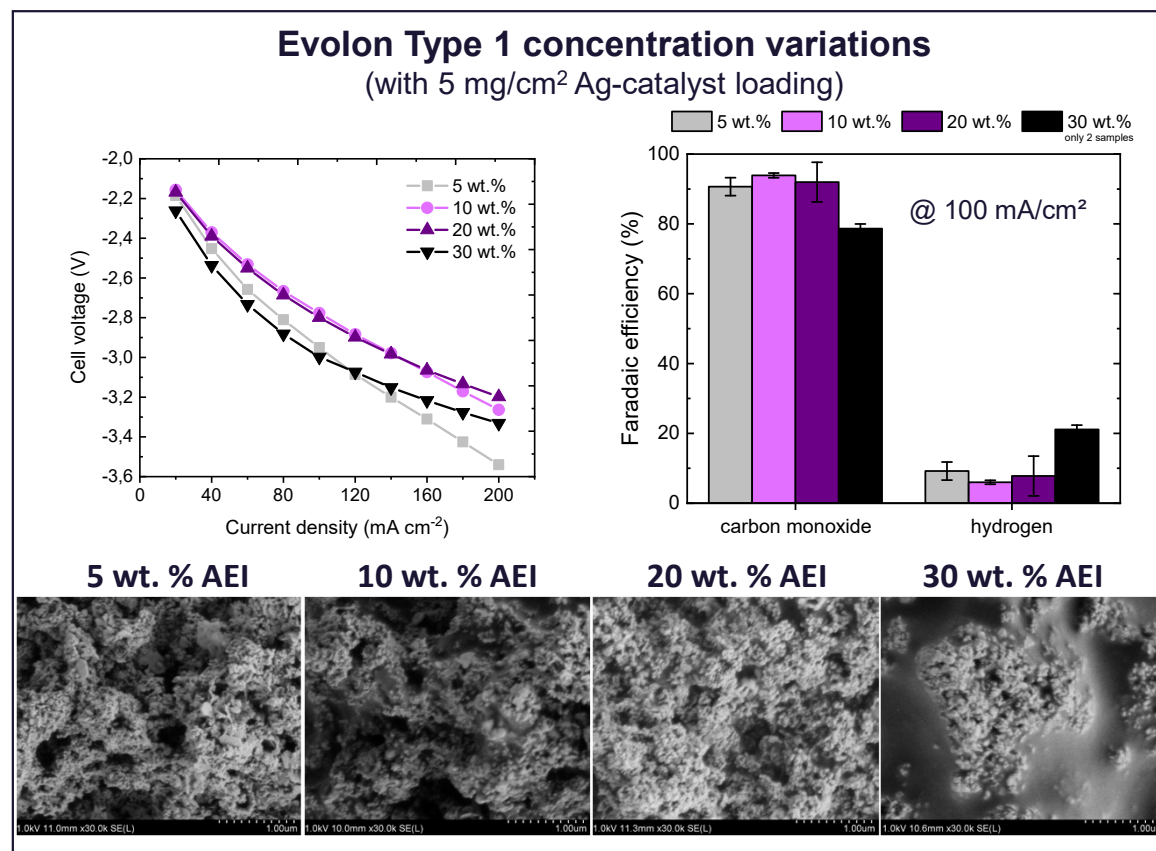
- Processing of the Ag-catalyst/AEI layers into GDEs using spray-coating.
- Investigation of electrochemical properties of the Ag-catalyst/AEI layers.

Result

- Testing procedure established.
- Cell setup is sensitive enough to display specific properties of different AEIs.

Outlook

- Evaluation of AEIs as prepared by Evonik and commercially available AEIs.
- Selection of best AEI for performance optimization and up-scaling.



Involved partners: Dr. A. Maljusch (Evonik); N. Martic, S. Szyszkowski, A. Maltenberger, A. Klinger, T. Reichbauer (SE)¹²
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WP4: GDE Characterization by impedance spectroscopy

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Objective

- Local impedance analysis



Approach

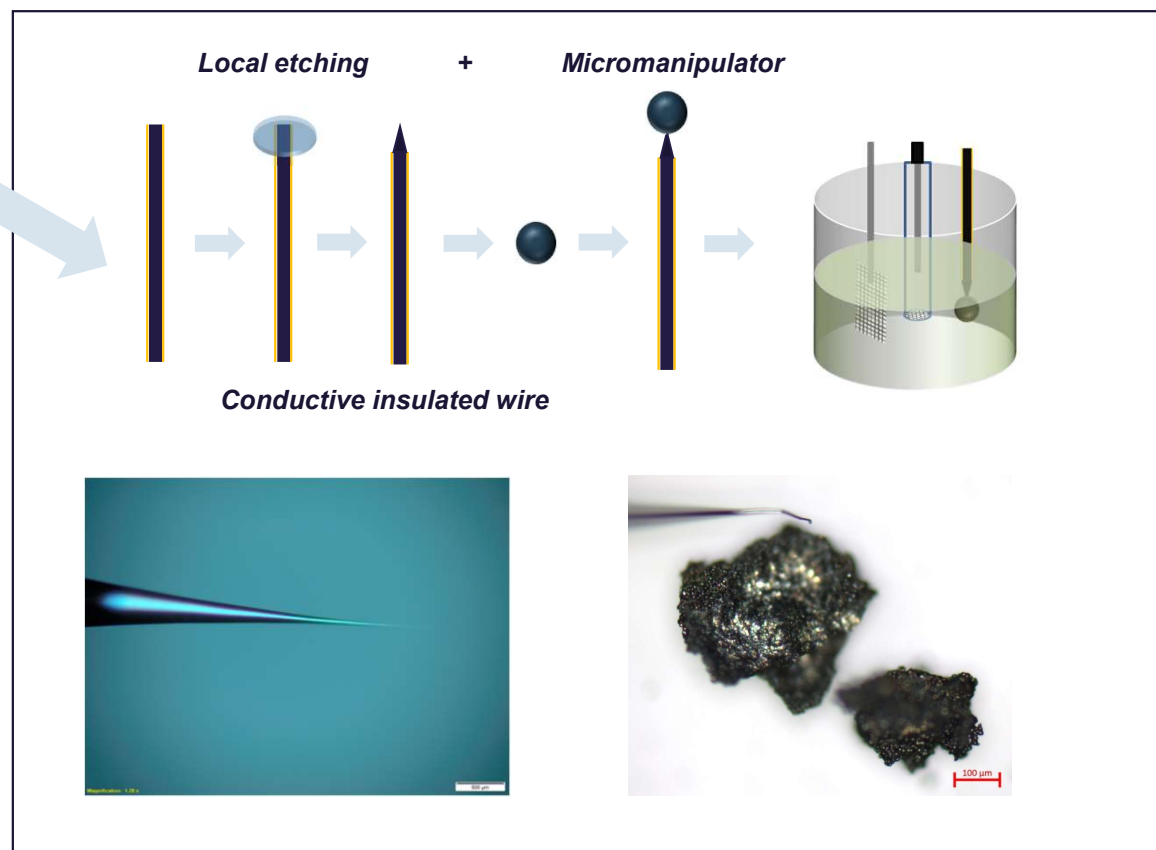
- Preparation of conductive tips
- Placing the particles at the tips and
- EIS investigations

Result

- The conductive tips were successfully produced by local electrochemical etching

Outlook

- Contacting the particles via inactive glues and performing EIS measurements



Involved partners: Dr. H. Tempel, Dr. Y. Emre, Dr. M. Schalenbach

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WP4: GDE Characterization by impedance spectroscopy

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Objective

- GDE characterization in the operating cell by impedance spectroscopy (EIS)

Approach

- Plane parallel anode (IrO_2) and cathode (GDE with catalyst). Reference electrode

Result

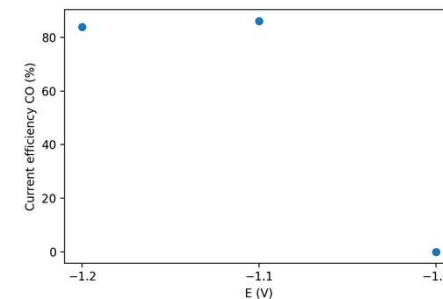
- Not possible to distinguish between hydrogen evolution and CO_2 reduction by EIS

Outlook

- Extract capacitance from impedance spectra and examine flooding/wetting

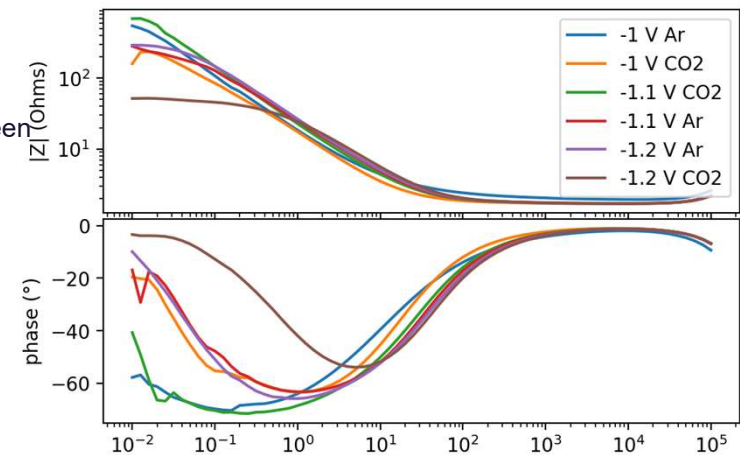
Current Efficiency:

Product composition varies as a function of the applied potential



Impedance spectra:

Not possible to distinguish between hydrogen evolution and CO_2 reduction, as all spectra can be characterized charge transfer resistance, capacitance and electrolyte resistance

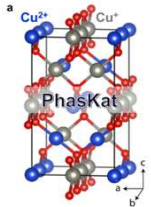


Involved partners: Dr. H. Tempel, Dr. Y. Emre, Dr. M. Schalenbach

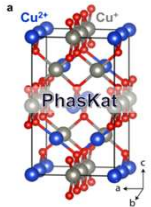
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Summary

- **WP1: Catalysts**
 - Synthesis accomplished for three catalyst materials.
 - Cu_4O_3 prepared by Evonik showing similar performance to reference samples.
- **WP2: Anion Exchange Ionomers**
 - First AEI samples of Evolon Type I (5g batch) prepared and evaluated.
 - Synthesis of Evolon scaled up to 10g batch.
- **WP3: Electrode Processing & Electrochemical Characterization**
 - Inks with higher IPA content contributes to stabilize the catalyst ink.
 - Inks with lower IPA content results more porous structure and better performance.
 - Cell setup and testing procedure for the evaluation of AEIs developed.
- **WP4: Impedance Spectroscopy**
 - Conductive electrodes with sharp tips were successfully prepared.
 - Not possible to distinguish between different reactions in the cell from impedance spectra.



Outlook



- Preparation of catalysts and AEIs and up-scaling of the synthesis ongoing.
- Continuing evaluation of catalysts and anion exchange materials.
- Selection of best catalyst and AEI for performance optimization and up-scaling.
- Correlating performance with ink composition and catalyst layer structure.
- The EIS analysis of the local electrode will be performed.
- Analyze wetting/flooding of the electrodes during operation in the cell with capacitance measurements.

Acknowledgement

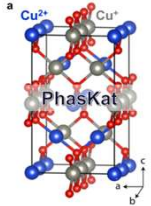
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Thank you for your attention!

Dr. Wiebke Sarfert-Gast

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